PUNCH

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1 English Cwt.

\$1.00 per U.S. Gallon

These data sheets are intended to be cut into four sections, 6 x 9 inches in size, as indicated by the straight lines. They may then be bound into note book form for convenient reference by means of staples inserted in holes punched at the points indicated. A suitable binder for these data sheets will be supplied for 25 cents. This binder has an open back and will hold an indefinite number of 6×9 sheets, depending on the length of staples used.

COMPARISONS OF MONEY STANDARDS.

Computed by Robert Grimshaw.

	Dollars.	Mai	ks. 1	Florin.	Franc.	Shilling.
\$ 1.00	1.00	4.2	0	2.46	5.18	4.12
Mk. 1.00		1.0	0	0.586	1.23	0.979
Fl. 1.00	0.406	1.7	1	1.00	2.10	1.67
Fr. 1.00	0.193	0.8	1	0.475	1.00	0.794
Sh. 1.00	0.243	1.0	21	0.599	1.26	1.00
\$1.00 per pound	Avoirdupois	=	5.41 11.40		per per ki	
1 sh. per pound	Avoirdupois '		$\frac{1.32}{2.77}$	Florins Francs	per per ki	ogram. kilogram. logram. kilogram.
\$1.00 per yard		=	4.59 2.69 5.66 4s. 66	Marks Florins Francs d. per 1	per M	leter.
1 shilling per ya	rd		1.38	Marks Florins Francs Dollars	per m	eter.
1 U. S. Gallon		=	3.79	liters =		3 British
1 British Gallon		=	452	liters =	= 1.2 U	J. S. gal.
1 Kilogram		=	2.2 po	unds av	oirdupo	is.
1 Pound Avoirdu	pois	=	0.453	kilogra	m.	
1 German Centn				U. S.		= 0.9843 wt.
1 U. S. Cwt.		=	0.885			= 45.36 $= 0.904$

1 shilling per British gallon	=	0.226	Marks per lite	r.
	=	0.133	Florins per li	ter.
	=	0.279	Francs per lit	er.
	=	0.054	Dollars per lit	er.
\$1.00 per U. S. Cwt.	=	4.63	Marks per Centner.	German
	=	2.71	Florins per Centner.	German
	=	5.71	Francs per Centner.	German
	=4	s. 6d	. 2 far. per Centner.	German
1 shilling per English Cwt.	=	1.00	Marks per Centner.	German
	=	0.59	Florins per Centner.	German

ALLOYS OF GREAT STRENGTH.

= 1.24

Thurston's composition: copper 55, zinc 44.5, tin .5. Composition used by government for consols of guns: copper 57.5, zinc 42, tin .5. One per cent. tin may be used, but it reduces the elongation. H. E. Field in Machinery, February, 1904, says: To produce good results the zinc and copper must be pure. All except the finest quality of zinc contains lead, which reduces strength. The purity of copper can be determined by testing its electrical conductivity; if the conductivity is high the copper is pure. In melting and particularly in pouring use as low a heat as possible without producing blowholes. High heating is very detrimental. It is advantageous to add a small amount of aluminum to check oxidation. When all new metal is used, melt copper first, cool, and add tin and zinc; stir with a carbon stick and when at correct temperature for pouring add the aluminum and pour at once. If scrap is used, melt copper first, add the scrap to reduce the temperature, then add tin and zinc.

Supplement to MACHINERY, April, 1904.

Francs per Centner.

Centner.

Dollars

German

German

LOGARITHMS OF CONVENIENT CONSTANTS.

German Centner.

= 1.12 U. S. Cwt, = 50.8 kilogram = 1.016 German Centner.

= 1.11 Marks per liter. = 0.65 Florins per liter. = 1.37 Francs per liter.

= 1s, 1d, 1 far. per liter.

Compiled by J. J. Clark.

	Logarithm	Reciprocal.	Logarithm
$\pi = 3.1416$.4971509	.318309	1.5028491
$\frac{\pi}{-} = .7854.\dots$	1.8950909	1.273237	.1049091
$ 4 $ $ \pi^2 = 9.86965 $.9943018	.10132	1.0056982
$\sqrt{\pi} = 1.772457$. 2485755	.5641888	$\overline{1.7514245}$
$\sqrt{\frac{1}{\pi}} = .564189$	1.7514245	1.772456	.2485755
$g = \stackrel{\pi}{32.16}$			2.4926840
$\frac{g}{2}$ $g = 16.08 \dots$			2.7937140
$\stackrel{?}{2}\stackrel{"}{g}=64.32\ldots$	1.8083460	.01554727	2.1916540
$\sqrt{2}g = 8.019974$			$\overline{1.0958270}$
1 cu. in. water weighs .03617	$ar{2}.5583485$	27.64723	1.4416515
Water column 1" x 1" x 1 weighs .43403 lbs		2.303988	.3624803
Water column 1"d x 1' weighs .34088 lbs		2.933584	.4673985
1 lb. water = column 1" x 1" x 2.304 '		.4340278	1.6375175
1 lb. water = column 1" d. x 2.9836'	. 4674009	. 340878	1.5325991
1 cu. ft. air at 32° F. and 30″ Hg weighs .08073 lbs	2.9070350	12.387	1.0929650
1 gal. H ₂ o weighs 8.355 lbs		.11969	$\overline{1.0780585}$
1 cu. ft. H2 o contains 7.48 gal.	.8739016	.13369	$\bar{1}.1260984$
14.7	1.1673173	.06802721	2.8326827
1728		.0005787037	$\overline{4}.7624563$
778		.001285347	$\overline{3}.1090204$
144	1		3.8416375
12			2.9208188
83000	4.5185139	.0000303	5.4814861
1 2			

COMPOSITION OF BRONZES (NAVY DEPARTMENT).

Contributed by F. W. Armes,

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	Parts.
Tin	
Copper	
Zinc	. 83.3
Antimony	. 3.8
Lead	. 3.0
Hard Bronze for Piston Rings:	
Tin	. 22.0
Copper	. 78.0
Bearings—Wearing Surfaces, etc.;	
Copper	. 6
Tin	. 1
Zinc	. 1/4
Naval Brass:	
Copper	62.0
Tin	
Zinc	37.0
Brazing Metal:	
Copper	85.0
Zinc	15.0
Antifriction Metal:	
Copper—(best refined)	3.7
Banca tin	88.8
Regulus of antimony	7.5
Well fluxed with borax and rosin in mixing.	
Bearing Metal—(Pennsylvania Railroad):	
Copper	77.0
Tin	8.0
Lead	15.0

BEARING METAL.

In the Journal of the Franklin Institute G. H. Clamer states that 13 parts antimony and 87 parts lead make an excellent bearing metal, these being exactly the proportions which give a homogeneous structure. For heavier duty tin should be added.

Supplement to MACHINERY, April, 1904

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